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### **Original Article**

### TERMINALIA TOMENTOSA ROXB (ex DC) WIGHT & ARN: PHYTOCHEMICAL INVESTIGATIO

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Date of Receipt-Date of Revision-Date of Acceptance- 24/07/2013

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#### ABSTRACT

The present study was undertaken to carry out phytochemical investigation of the ethanolic extract of the stem bark of Terminalia tomentosa Roxb (ex DC) Wight & Arnbelonging to the family Combreteaceae. The plant is known in Sanskrit as Asana, in English as Black murdah, in Hindi as Asan, Saj, Sain and in Marathi as Ain. The plant has been known to possess various pharmacological antioxidant, activities like antifungal, antihyperglycaemic, antidiarrhoeal, antileucorrheal etc. The bark of the plant is astringent& useful in ulcers, vata, fractures, haemorrhages, bronchitis, diarrhoea etc.Preliminary phytochemical screening of the ethanolic extract of stem bark revealed the presence of carbohydrates, flavonoids, triterpenoids, steroids, tannins and saponins. The chemical entities isolated and characterised includes 4 - methy 1 - 4 hydroxymethylene -  $6\beta$  - (10 - methyl octanyl) cyclohexane (Arjuna homoses quiterpenol), di-n-octyl phthalate, di isobutyl phthalate and dibutyl phthalate. All these phytoconstituents are reported for the first time from the ethanolic extract of stem bark of T.tomentosa.

**Keywords**: Terminalia tomentosa,Combreteaceae, Di-n-octyl phthalate, Triterpenoids.

#### **INTRODUCTION**

*Terminalia tomentosa* Roxb (ex DC) Wight & Arn (Synonyms: *Terminalia alata* Heyne ex Roth, *Terminalia crenulata Roth*, *Terminalia ellipticaWilld.*) is a large deciduous tree, 20-35m high & 1m in diameterbelonging to family Combretaceae<sup>1,2,3,4</sup>. The bark is rough, dark grey to black in colour with deep vertical fissures & transverse cracks <sup>[5]</sup>. Leaves are simple, sub-opposite or the uppermost alternate, thick coriaceous, ovate-oblong or elliptic-oblong, rarely obovate, softly tomentose when young ;becoming more or less glabrous when mature, with 1-2 glands (

**American Journal of Advanced Drug Delivery** 

which are often turbinate or long stalked ) usually on the midrib but sometimes absent. Flowers are hermaphrodite and in axillary fulvous-pubescent spikes or terminal panicles. Fruits are  $1^{1}/2$  -2 inches long and <sup>3</sup>/<sub>4</sub> inch wide with 5 broad, coriaceous, brown, glabrous wings striated with numerous straight lines running horizontally from the axis to the edges<sup>[6]</sup>. The plant is common in the forests, especially in the humid regions of India, including the sub-Himalayan tracts of North West provinces, Nepal & Sikkim. also Southwards throughout the Peninsula <sup>7</sup>. It is a prominent part of both dry and moist deciduous forests in southern India up to 1000 m. The bark is bitter & stypic, useful in vitiated conditions pitta, ulcers. vata. of fractures, haemorrhages, bronchitis cardiopathy. strangury, wounds, haemoptysis, dysentery, cough, verminosis ,leucorrhoea, gonorrhoea burning sensation  $(Ayurveda)^{[8,9]}$ . & Phytoconstituents such as tannins like arjunic acid, arjunolic acid, arjunetin, ellagic acid, gallic acid, and triterpenoids like oleanolic acid, betulinic acid and steroid like  $\beta$ -sitosterol have been reported to be present in *T.tomentosa*<sup>10, 11, 12, 13, 14</sup>. The plant is known to possess many pharmacological properties like antifungal <sup>15</sup>, antioxidant <sup>[16]</sup>, anti-hyperglycaemic<sup>17</sup>, anti-diarrhoeal& anti leucorrheal<sup>18</sup>. From the literature survey, it was learnt that no substantial work has been carried out on the stem bark of T.tomentosa .Hence an attempt was made to investigate the phytoconstituents from ethanolic extract of *T.tomentosa*stem bark.

#### **MATERIALS AND METHODS**

All the melting points were recorded in Bio Technics India, Model no.BT2-38 melting point apparatus & were uncorrected. IR spectra of the compounds were recorded using the KBr pellet method on a Bruker  $\alpha$ -T Spectrophotometer, at National facility for Clinical Trial, ISISM Chennai.<sup>1</sup>HNMR & LC-MS spectra of compounds were taken Bruker 500 MHz PMR on Spectrophotometer using CDCL<sub>3</sub> as solvent & Shimadzu LC 2020 at National facility for Clinical Trial, ISISM Chennai, ESI-MS spectra were recorded using ESI-MS Expression CMS Advion at SvnZeal Research Laboratory, Ahmedabad. TLC was carried out using Aluchrosep Silica gel 60/UV254 from S.D. Fine Chemicals Pvt. Ltd, Mumbai. Column chromatography was carried out using glass column with a glass stop cock. 30×600mm from Merck Specialities Pvt. Ltd, Mumbai, packed with silica gel (200-400 mesh) from Molychem, Mumbai. All the chemicals & reagents used were obtained in high purity from S.D. Fine chemicals Pvt. Ltd, Bombay, Molychem & Chemport Pvt. Ltd, Mumbai.

#### Plant Material<sup>6</sup>

The stem bark of *T. tomentosa* was collected from Darbandora, Ponda- Goa during October 2012. It was authenticated by Prof G. I. Hukkeri, Dept. of Botany, Dhempe College of Arts & Science, Miramar-Goa.

The stem bark was then washed thoroughly to remove the soil and adhering materials and dried in shade. The dried stem bark was powdered and used for the preparation of ethanolic extract.

#### Preparation of ethanolic extract<sup>19</sup>

The dried stem barkpowder (500g) was extracted by maceration with ethanol (95%) for 3 days. After 3 days ethanolic layer was decanted off. The process was repeated thrice. The solvent from the total extract was distilled off using Rotary vacuum evaporator (Superfit) and the concentrate was evaporated to a syrupy consistency and then evaporated to dryness (80g)

# Preliminary Phytochemical Screening (Qualitative Analysis)<sup>20,21</sup>

The preliminary phytochemical studies were performed for testing the different phytoconstituents present in the ethanolic extract of the stem bark of *T.tomentosa*as per the standard procedures. The results are tabulated in table 1.

## Isolation of Compounds from Ethanol Soluble Fraction<sup>19</sup>

The ethanol soluble fraction (10g) was mixed with silica gel (2g). The sample was then loaded on columnpreviously packed with 150g of Silica gel (Molychem, 200-400mesh ) prepared in petroleum ether (60-80°C). The column was subjected to different solvent systems, starting first with petroleum ether 100% followed bv petroleum ether:chloroform graded mixtures (95:5,90:10,80:20,70:30,60:40,50:50) then with chloroform 100% followed by graded mixtures of chloroform : ethyl acetate (95:5,90:10,80:20,70:30,60:40,50:50) & finally with ethyl acetate 100% & graded mixtures of ethyl acetate: methanol (99:1,98:2,97:3,96:4,95:5). The elutions were monitored by TLC (Silica gel-G), visualization by UV 254,366nm & vanillinsulphuric acid spraying reagent heated at 110°C.Each time 10ml elutes were collected & identical elutes were combined (TLC monitored) & concentrated to 5ml & kept aside.

Elutions carried out with graded mixture of petroleum ether (60-80°C): (80:20) chloroform, resulted a single component on TLC (petroleum ether (60-80°C): chloroform, 80:20). After removal of the solvent, an off white powder was obtained, which was designated as **Compound AM 1** (70mg).

Elutions carried out withethyl acetate:methanol (99:1) resulted into a single component on TLC (ethyl acetate:methanol, 99:1). After removing

solvent yellow liquid resulted, this was designated as **Compound AM 2** (62mg).

Elutions carried out with ethyl acetate:methanol (97:3) resulted into a single component on TLC (ethyl acetate:methanol, 97:3). After removing solvent yellow viscous liquid resulted, this was designated as **Compound AM 3** (55 mg).

Elutions carried out with ethyl acetate:methanol (95:5) resultedinto a single component on TLC (ethyl acetate:methanol, 95:5). After removing solvent pale yellow viscous liquid resulted, this was designated as **Compound AM 4** (45 mg).

Elutions carried out with other graded mixtures of solvents resulted in resinous mass which was not processed further.

#### **RESULTS AND DISCUSSION**

The preliminary phytochemical screening of the ethanolic extract of stem bark of T.tomentosa was performed and it was found to contain carbohydrates, flavonoids, triterpenoids, steroids, tannins and saponins. The chemical investigation led to the isolation of four compounds from the ethanolic extract of the stem bark of T.tomentosa. The isolated compounds were  $4 - \text{methyl} - 4 - \text{hydroxymethylene} - 6\beta$  -(10 - methyl octanyl) cyclohexane (Arjunahomosesquiterpenol), di-n-octyl phthalate, di isobutyl phthalateand dibutyl phthalate.

Compound 1 (Arjuna homoses quiterpenol): m.p.152oC; IR (KBr): 3437.39 cm<sup>-1</sup> (br, OH), 2925.14 cm<sup>-1</sup> (C-H str. in CH3), 2858.08 cm<sup>-1</sup> (C-H str. in CH2), 1458.11 cm<sup>-1</sup> (C-H deformation in CH3), 1279.29 cm-1 (C-O str.); 1HNMR (CDCl3): $\delta$  0.779 –  $\delta$  0.956 (m, 3H, H-14),  $\delta$ 0.977 (s,3H, H-17),  $\delta$  1.078 (brs, 3H, H-16),  $\delta$  1.258 (s, 12H, 6xCH2),  $\delta$  1.511- $\delta$ 1.550 (m, 4H, H-2, H-5 of the ring),  $\delta$  1.557

(s,4H, H-1, H-3 of the ring) ,δ 2.314-δ 2.393 (m,1H, H-10α), δ 2.413-δ 2.592 (m, 1H, H-6α) ,δ 3.730-δ 3.815 (m,1H, H2-15b) ,δ 3.828 (s,1H,OH), 84.035-84.089 (m,1H,H2-15a):In the LCMS spectrum of Arjuna homoses quiterpenol, it exhibited molecular ion peak at m/z 255.10  $[M+H]^+$  which was consistent with molecular formula of C17H34O. Fragment ion peaks at m/z 239.20  $[M-Me]^+$ , 225.10  $[M-C2H5]^+$ , 223.25 [M-CH2OH]<sup>+</sup>, 211.45 [M-C3H7]<sup>+</sup>, 197.10 [M-C4H7], 169.45 [M-C6H13]<sup>+</sup>, 155.30 [M-C7H15]<sup>+</sup>, 141.25 [M-C8H17]<sup>+</sup>, 127.25 [M-C9H19]<sup>+</sup> suggested that the molecule possessed a C9-side chain attached to a hydroxyl substituted dimethvl ( cyclohexane ring $)^{22}$ . (Figure 1).

Compound 2 (Di-n-octyl phthalate): b.p. 3790C; IR (KBr): 2927.17 cm<sup>-1</sup> (C-H str. in CH3), 2860.76 cm-1 (C-H str. in CH2), 1727.64 cm<sup>-1</sup> (C=O str.), 1628.74 cm-(C=C str.), 1459.24  $cm^{-1}$ (C-H 1 deformation in CH3), 744.07 cm<sup>-1</sup> (C-H bending of aromatic ring); 1HNMR (CDCl3): 8 0.866- 8 0.997 (m, 10H, H-7', 7", 8', 8"),δ 1.364- δ 1.439 (m, 12H, H-4',4", 5', 5",6',6"),δ 1.674- δ 1.747 (m, 8H, H-2', 2", 3', 3"), 8 4.184- 8 4.256 (m, 4H, H-1', 1"),δ 7.510- δ 7.535 (m, 2H, H-4,5), δ 7.697- δ 7.723 (m, 2H, H-3,6); The ESI-MS spectrum showed molecular ion peak at m/z 391.4  $[M+H]^+$  in the positive ion mode which was consistent with molecular formula of C24H38O4.Fragment ion peaks at m/z 279.2, m/z167.1 exhibited in the mass spectrum were characteristic of alkyl phthalates & the base peak at m/z 149 is due to the protonated phthalic anhydride  $(C8H5O3)^{23}$ .(Figure 2).

Compound 3 (Di isobutyl phthalate): b.p.  $319^{\circ}$ C; IR (KBr): 2928.50cm<sup>-1</sup> (C-H str. in CH<sub>3</sub>), 2861.96cm<sup>-1</sup> (C-H str. in CH<sub>2</sub>),1727.99cm<sup>-1</sup> (C=O str.),1603.24cm<sup>-1</sup> (C=C str.),1457.21cm<sup>-1</sup> (C-H deformation in CH<sub>3</sub>),1279.29cm<sup>-1</sup> (C-O str.),1125.66cm<sup>-1</sup> (C-H deformation in CH<sub>2</sub>), 830.36cm<sup>-1</sup> (C-H

<sup>1</sup>HNMR bending of aromatic ring); (CDCl<sub>3</sub>): 8 0.863-8 1.684 (m, 12H, H-3', 3", 4', 4"),δ 3.574- δ3.621 (m, 2H,H-2',2"), δ 4.202- 84.319 (m, 4H, H- 1',1"), 8 7.513δ7.537 (m, 2H,H-3,4), δ 7.693- δ7.723 (m, 2H,H-2,5); The ESI-MS spectra showed molecular ion peak at  $m/z 278.3 [M]^+$  which was consistent with molecular formula of C<sub>16</sub>H<sub>22</sub>O<sub>4</sub>. The mass spectrum showed fragment ion peaks at m/z 167.1 & m/z 149 which are considered characteristic of alkyl phthalates & m/z 223.2 suggested that the alkyl phthalate is Di isobutyl phthalate. The ion peak at m/z 149 is due to the protonated phthalic anhydride  $(C_8H_5O_3)^{24}$ . (Figure 3).

Compound 4 (Dibutyl phthalate): b.p.337°C; IR (KBr):2925.79 cm<sup>-1</sup> (C-H str. CH<sub>3</sub>),2858.25  $cm^{-1}$ (C-H str.in in CH<sub>2</sub>),1728.23 cm<sup>-1</sup> (C=O str.), 1637.65 cm<sup>-1</sup> (C=C str.), 1458.18 cm<sup>-1</sup> (C-H deformation in CH<sub>3</sub>), 1073.39 cm<sup>-1</sup> (C-O str.), 742.94 cm<sup>-1</sup> (C-H bending of aromatic ring); <sup>1</sup>HNMR (CDCl<sub>3</sub>):  $\delta$  0.896-  $\delta$  0.998 (m, 6H, H-4', 4"), δ 1.257- δ 1.424 (m, 4H, H-3',3"), δ 1.719 (s,4H, H-2',2"), δ 4.184- δ 4.320 (m,4H,H-1',1"), δ 7.505- δ 7.543 (m, 2H, H-4,5),δ 7.689- δ 7.723 (m,2H,H-3,6); The ESI-MS spectrum showed molecular ion peak at m/z 279.2  $[M+H]^+$  which was consistent with molecular formula  $C_{16}H_{22}O_4$ . The mass spectrum showed fragment ion peaks at m/z 149which 167.1&m/z are considered characteristic of alkyl phthalates & used in their characterization. The fragment ions at m/z 205.4 & 223.2 suggested that the alkyl phthalate is Dibutyl phthalate. The base peak at m/z 149 is due to the protonated phthalic anhydride  $(C_8H_5O_3)^{24}$ . (Figure 4).

#### CONCLUSION

The chemical investigation led to the isolation of four compounds from the ethanolic extract of the stem bark of *T.tomentosa*, which includes 4-methyl – 4 – hydroxymethylene -  $6\beta$  - (10 -methyl octanyl) cyclohexane (Arjunahomosesquiterpenol), di-

n-octyl phthalate, di isobutyl phthalateand dibutyl phthalate. The isolated and characterised constituents can be categorised under the class of sesquiterpenoids and phthalate derivatives. The above compounds were isolated for the first time from theethanolic extract ofstem bark of T.tomentosa.

#### ACKNOWLEDGEMENTS

The authors are grateful to the Authorities of Government of Goa and the Principal, Goa College of Pharmacy for their immense support and providing the laboratory facilities. Authors are also thankful tothe Principal and staff members, National Facility for Clinical Trial, Interdisciplinary School of Indian System of Medicine, Chennai; and SynZeal Research Laboratory, Ahmedabad for providing the spectras. We are also grateful to Prof. G.I. Hukkeri, Dept. of Botany, Dhempe College of Arts and Science, Miramar-Goa for authenticating the plant material.

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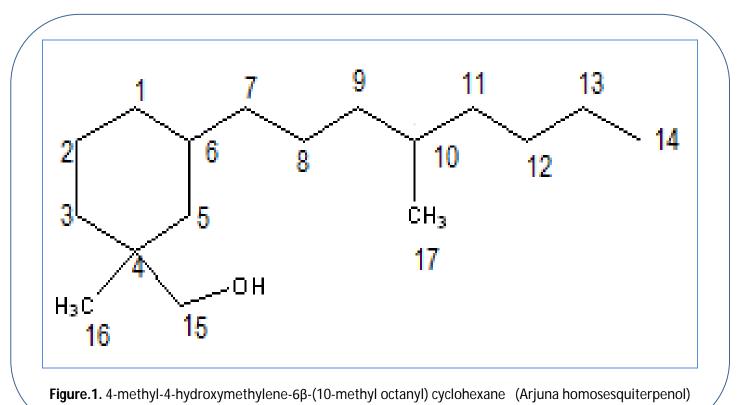
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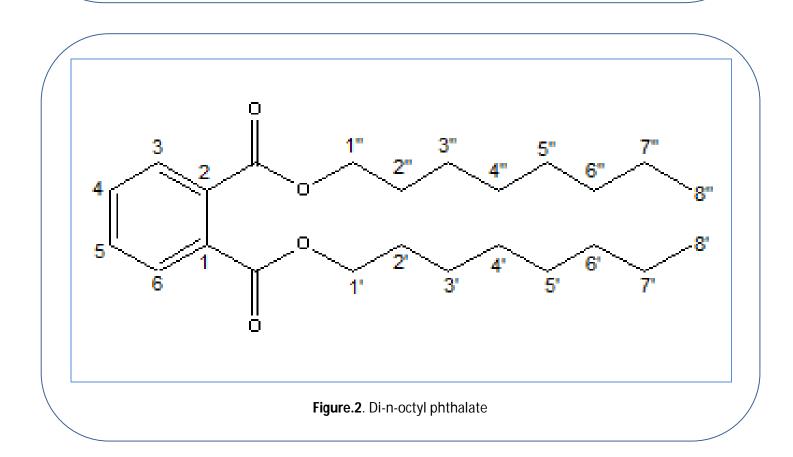
**Table 1.** Result of Qualitative analysis for phyto-constituents isolated from the ethanolic extractof the stem bark of *T.tomentosa* W& A.

SR.NO	CHEMICAL TEST	INFERENCE
1.	Alkaloids	-
2.	Carbohydrates	+
3.	Flavonoids	+
4.	Triterpenoids	+
5.	Steroids	+
6.	Tannins	+
7.	Resins	-
8.	Proteins	-
9.	Glycosides	-
10.	Starch	-

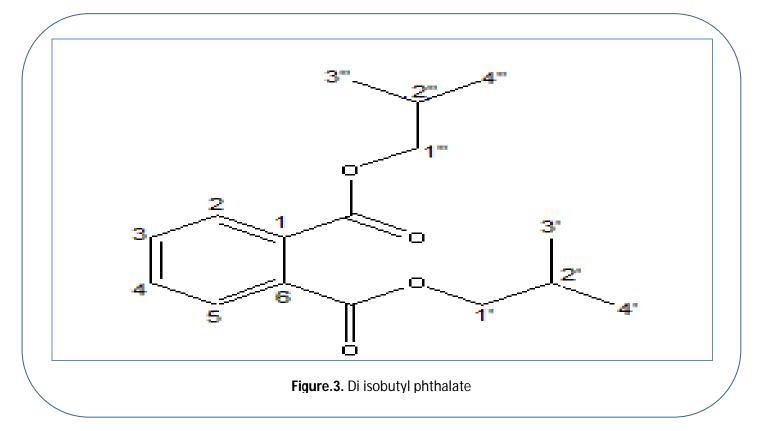
Present = + Absent = -

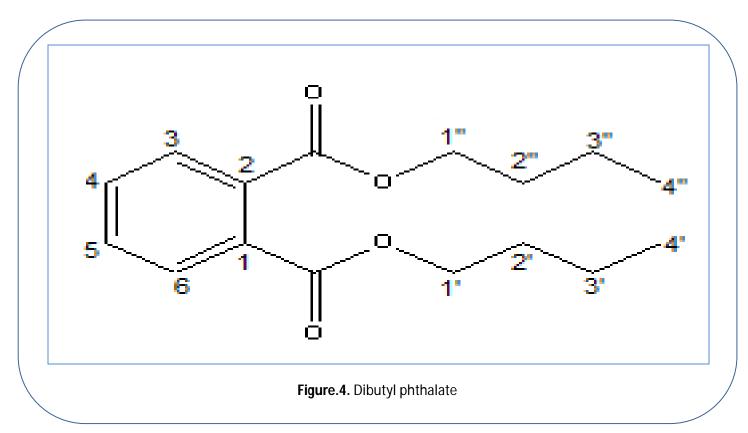






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