# The use of hydrogen-deuterium exchange to investigate the role of structural conformation and protein folding of the amyloidogenic protein b<sub>2</sub>-microglobulin.



# Overview

Studies into the conformational dynamics of  $\beta_2$ -microglobulin ( $\beta_2$ m) using hydrogen deuterium exchange (HDX) coupled to electrospray ionisation mass spectrometry (ESI-MS)<sup>1</sup> have been initiated. A comparison of the rate of exchange with variations in pH and temperature has been made.

# Introduction

 $\beta_2$ -microglobulin ( $\beta_2$ m) is an 11,860.4 Da protein comprising seven  $\beta$ -sheet strands (Figure 1). In its physiological role,  $\beta_2$ m forms the light chain of the major histocompatibility complex and is important in cellular immunity. However it is also the building block of debilitating amyloid deposits which are found in the joints of dialysis-related amyloidosis sufferers<sup>2</sup>.

Catabolism of free  $\beta_2 m$  by the kidney normally regulates the blood concentration of the protein. However in kidney failure this mechanism is lost and  $\beta_2$ m concentrations in the blood can increase 60-fold. Dialysis does not remove  $\beta_2$ m and over the course of years  $\beta_2$ m monomers aggregate to form tissue specific amyloid fibrils. Clinical symptoms of DRA are seen in 100% of dialysis patients after 15 years of treatment.



The mechanism by which the amyloid fibrils are formed is not known nor is the structure of the fibrils. Our previous ESI-MS work has provided interesting insights into the conformational properties of  $\beta_2 m^{3,4}$ . Here we are using HDX to investigate the conformational dynamics of the protein.

## Methods

 $\beta_2 m$  was deuterated by dissolving in  $D_2O$  (5 mg mL<sup>-1</sup>); after 7 days the solution was lyophilised and diluted in D<sub>2</sub>O to 5 mg mL<sup>-1</sup> (Figure 2). HDX was initiated by a 1:25 dilution of this solution into a 5mM ammonium acetate/ 5 mM ammonium formate buffer adjusted to the required pH with HCl or ammonium hydroxide. Aliquots were removed at time intervals and analysed by ESI-MS.

ESI-MS on the Platform II (Waters Corp., Manchester, UK) used a flow rate of 10  $\mu$ L min<sup>-1</sup>, a capillary voltage of 3.5 kV and a cone voltage of 40V. Nano-ESI-MS on the LCT Premier (Waters Corp., Manchester, UK) was achieved with a NanoMate temperature controlled interface (Advion, Ithaca, NY, USA) using a spray voltage of 1.7 kDa and a cone voltage of 40V.

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### Results

The extent of  $\beta_2$ m deuteration was measured by ESI-MS (Figure 2). The mass difference of 197 Da indicates complete deuteration.



**Figure 2: ESI-MS m/z spectra of deuterated (lower)** and non-deuterated  $b_2m$  (upper) acquired at pH 7.

HDX of deuterated  $\beta_2$ m was carried out at pH 5 at 4°C, and at pH 7 at 4°c and 22°C. The reaction was initiated, temperature controlled and sampled using the NanoMate interface on the LCT Premier (Figure 3). At 2 mins, the fastest exchange was at pH 7 (22°C) where 140 deuteriums had been replaced with hydrogens (Table 1). By 60 mins, all three reactions indicated a protected core of ~35 deuterated sites. This is consistent with EX1 kinetics, while peaks due to both EX1 and EX2 kinetics can be seen between 3 and 5 hours (Figure 4).



**Figure 3:** HDX of b<sub>2</sub>m monitored by nanoESI-MS at time: (i) 2 mins; (ii) 20 mins; (iii) 40 mins; (iv) 60 mins.

### Figure 1: b<sub>2</sub>m

Time (mins)	pH 7 4°C	pH 7 22°C	рН 5 4°С
60	-162	-163	-163
30	-161	-161	-160
20	-141	-157	-151
2	-131	-140	-120



### Conclusion

NanoESI-MS with the NanoMate can be used effectively for temperature controlled HDX. Further work planned includes confirming the EX1/EX2 exchange kinetics of  $\beta_2$ m HDX mechanisms and studying the HDX properties of a series of  $\beta_2$ m variants with different amyloidogenic properties.

### References

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### **Table 1: HDX of** b<sub>2</sub>**m monitored by nano-ESI-MS** showing loss of deuterium with time

### **Figure 4:** Mass profile time course of b<sub>2</sub>m HDX at pH 5 monitored by ESI-MS. The reaction was cooled to 4°C for the first minute to slow the initial rate and then continued at 22°C.

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